

## NOTE

L-3-(2,5-DIHYDROPHENYL)ALANINE,  
AN ANTIMETABOLITE OF  
L-PHENYLALANINE PRODUCED  
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An unclassified species of a Streptomyces, X-13,185, when grown in a medium containing glucose, 10 g/liter; asparagine, 0.5 g/liter; yeast extract (Difco), 2 g/liter and  $K_2HPO_4$ , 0.5 g/liter produced a substance (100 mg/liter at 48 hours) inhibitory to the growth of *Bacillus subtilis* and *Escherichia coli* on DAVIS and MINGIOLI minimal agar<sup>1</sup>. The growth inhibition was relieved by addition of L-phenylalanine to the agar.

The antimetabolite was isolated from the filtered broth by adsorption onto Dowex 50WX4 resin (50~100 mesh) in the hydrogen ion form. The activity was then eluted with a 5% aqueous solution of pyridine. After removal of the pyridine by evaporation at reduced pressure the antimetabolite was adsorbed onto Norite A at pH 6.5 and eluted from the charcoal with 50% aqueous ethanol. The antimetabolite was then separated from aromatic amino acids by cation-exchange chromatography. A solution containing 5 g of crude material at pH 2.3 was adsorbed onto a 57 cm x 5 cm column containing AG 50WX4 resin (200~400 mesh) in the sodium ion form and then eluted with 0.4 M sodium citrate buffer at pH 4.2. The active fractions free of phenylalanine were desalted by adsorption onto AG 50W x 4 resin (50~100 mesh) in the hydrogen ion form followed by elution with 5% aqueous pyridine solution. After evaporation at reduced pressure, the pH was adjusted to 0.8 with 5 N hydrochloric acid and the hydrochloride salt was

crystallized from ethanol. Recrystallization from ethanol resulted in colorless needles of L-3-(2,5-dihydrophenyl) alanine hydrochloride m. p. 206°~208°C;  $[\alpha]_D^{25} -33.7$  (c 1, 5 N HCl)  $-47.9$  (c 1, 0.2 N sodium phosphate pH 6.5). Anal. calcd. for  $C_9H_{13}NO_2 \cdot HCl$ : C 53.08, H 6.93, N 6.88, Cl 17.60. Found: C 52.99, H 7.03, N 6.85, Cl 17.52. When crystallized as the zwitterion a rapid partial conversion of the anhydrous crystals to phenylalanine occurred on exposure to air.

The oxidative conversion to phenylalanine and the elemental analysis suggested a dihydrophenylalanine. The mass spectrum of the zwitterion with peaks at m/e 167, 166, 165, 122, 120, 93, 91, 74 contained the molecular ion, m/e 167, and the M-45 (loss of COOH), and M-74 (loss of  $H_2NCHCOOH$ ) peaks. In addition a series of peaks two mass units lower at m/e 165, 120 and 91 corresponds to the mass spectrum of phenylalanine. The lack of ultraviolet absorption at wave lengths higher than 230 nm in samples free of phenylalanine excluded conjugated double bonds and the nmr spectrum of the zwitterion ( $D_2O$ , ext. TMS) pointed to a 2,5 rather than a 1,4 dihydro system since there were three olefinic protons,  $\delta$  6.19(s, 2), 6.12(s, 1) and four allylic protons,  $\delta$  3.12. Confirmation was obtained by comparison of the properties with authentic<sup>2,3</sup> material synthesized by reduction of L-phenylalanine with lithium in ammonia.

Since L-3-(2,5-dihydrophenyl)alanine is new in nature we carried out a preliminary investigation into its biosynthesis. No incorporation of label from L-phenylalanine- $C^{14}$  (U) at a detection limit of 0.02% was observed. Label from shikimic acid- $C^{14}$  (U) was incorporated into L-3-(2,5-dihydrophenyl)alanine to the extent of 2.5%. Furthermore, over 90% of the label in the extracellular amino acid pool was present in L-3-(2,5-dihydrophenyl)alanine. Presumably, biosynthesis is accomplished by way of a branch in the aromatic amino acid pathway.

A report<sup>4</sup>) on the microbiological activity of L-3-(2,5-dihydrophenyl)alanine has appeared. In addition, we have observed

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activity against sarcoma-180 in the mouse when administered intraperitoneally and against *Trichomonas vaginalis* when administered locally.

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#### References

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